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Chemistry Benchmark Study Guide Answers

Chemistry Benchmark Study Guide Study your notes from the following lessons: 1. How would one distinguish between the subatomic particles that make up an atom? 2. How are elements identified using the periodic table? 3. How are the states of matter determined? 4. How are chemical changes distinguished from physical changes? 5.

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Chemistry Study Guide Answers. 1. Using the periodic table, how many protons, neutrons, and electrons are in an atom of cobalt? $P = 27E = 27N = 32$. *Electrons = protons, protons = atomic number, neutrons = atomic mass - protons.

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