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A sealed vessel contains 50% oxygen, 10% carbon dioxide, and 40% nitrogen gas. The total pressure of the gas mixture is 5 atmospheres. What is the partial pressure of the carbon dioxide?

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States that equal volumes of gases at the same temperature and pressure contain equal numbers of molecules. Standard Molar Volume of a Gas The volume occupied by one mole of a gas at STP. This is found to be 22.4 L.

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